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# Behavior Of Gases Webquest Answer Key

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Worksheet atm =  
760.0 mm Hg = 101.3  
kPa= 760 .0 torr  
Boyle's Law Problems:  
1. If 22.5 L of nitrogen  
at 748 mm Hg are  
compressed to 725  
mm Hg at constant  
temperature. Gas Law  
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DIRECTIONS. On this  
quest, you will be  
working in teams of 2,  
gathering information  
from each website  
listed. There are 2 jobs  
in each team: the  
navigator and the data  
recorder. Navigator – is  
responsible for logging  
in, navigating the  
Internet – using the  
touch pad and  
maneuvering around  
the websites, and  
logging out. Data  
Recorder – records all  
data NEATLY in your  
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+ Gaseous State 07 +

Real Gas and Ideal Gas

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data NEATLY in your packet (teammate will copy answers later).

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molecules are far apart  
so volume of gas  
molecules are  
negligible and...